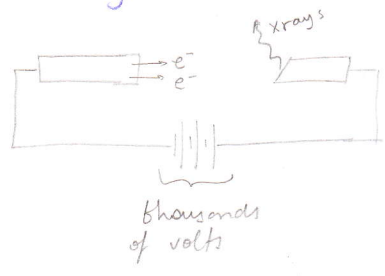


X-Rays

W. K. Roentgen (1895)



X rays

- no deflection in \vec{B}
- was not able to see diffraction, interference
- most materials - transparent to it
 - decreases with material density

found medical application (achieve photographic film)

hard to see diffraction (bending of waves caused by obstacles) because $\lambda = 1 \text{ \AA}$

↳ need slit $\sim 1 \text{ \AA}$

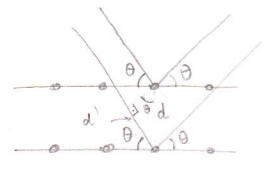
↳ crystal (array of atoms separated $\sim 1 \text{ \AA}$) \Rightarrow diffraction

confirmed that X rays are electromagnetic waves

Bragg (1912)

↳ use of X rays to determine the structure of crystals

- treat atomic planes as reflecting mirrors
- assume \angle of incidence = \angle of reflection



$$\sin \theta = \frac{d'}{d}$$

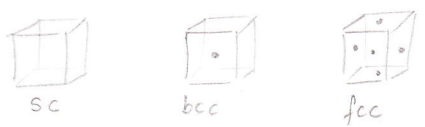
path difference: $\Delta = 2d' = 2d \sin \theta$

interference is constructive if $\Delta = n\lambda$ $n = 1, 2, 3, \dots$

$2d \sin \theta = n\lambda$

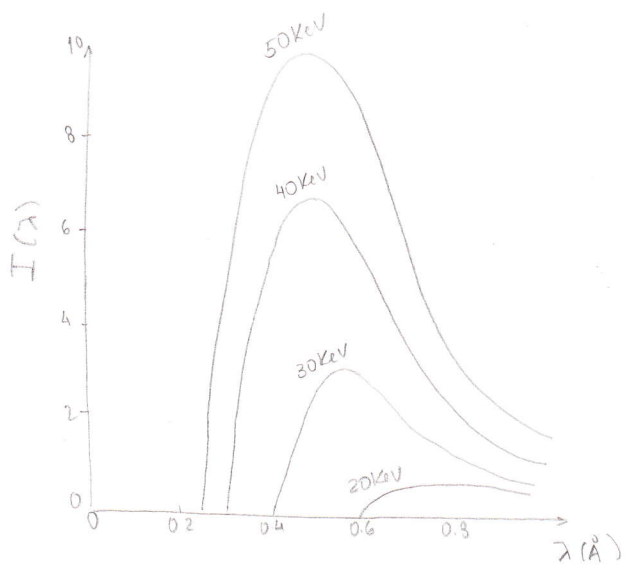
Bragg's law

λ is known, θ from experiment \Rightarrow can obtain the interplanar distance d



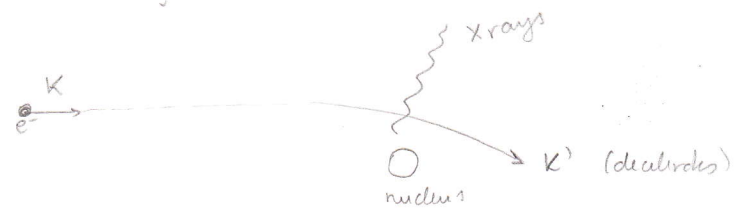
Study of the X-ray spectrum

↳ depends on the accelerating voltage and material



(1) continuous X radiation

bremstrahlung
decelerating radiation



(2) sharp lines (NOT shown)

(3) cutoff λ_{min}

↳ indep of material

dep on energy of e^- / the potential

$$h\nu = K - K'$$

{ an inverse photoelectric effect

↓ photon absorbed / E, p go to e^-

$$\frac{hc}{\lambda} = K - K'$$

bremstrahlung: photon created / E, p from e^- colliding with nucleus

$$K' = 0 \Rightarrow \boxed{K = \frac{hc}{\lambda_{min}}}$$

(electron loses all its kinetic energy)

$$K = eV \Rightarrow eV = \frac{hc}{\lambda_{min}} \Rightarrow \boxed{\lambda_{min} = \frac{hc}{e} \frac{1}{V}}$$

voltage

Exercise: Determine h from the fact that $\lambda_{min} = 3.11 \times 10^{-11} m$ for 400 KeV electrons.

$$K = h\nu = \frac{hc}{\lambda} \Rightarrow h = \frac{K\lambda}{c}$$

$$h = \frac{400 \times 10^3 eV \times 1.6 \times 10^{-19} J \times 3.11 \times 10^{-11} m}{3 \times 10^8 m/s}$$

$$= \boxed{6.64 \times 10^{-34} J \cdot s}$$

$$\frac{1240 (eV) (nm)}{1.60 \times 10^{-19} C} \Rightarrow \frac{1240 \times 1.60 \times 10^{-19} J}{1.60 \times 10^{-19} C} (nm) = 1.24 \times 10^3 V \cdot nm$$

unit